

LAMPIRAN A

CONTOH PERHITUNGAN PERSENTASE HASIL SINTESIS

I. Perhitungan berat teoritis

a. Ammonium tiosianat (BM : 76,12 g/mol)

Penimbangan : 1,14 gram

$$\text{mmol ammonium tiosianat} : \frac{1,14 \times 1000}{76,12} = 14,98 \sim 15$$

b. Benzoil klorida (BM : 140,57 g/mol, berat jenis : 1,2070 g/cm³)

Volume : 1,2 ml

$$\text{mmol benzoil klorida} : \frac{1,2 \times 1,2070}{140,57} \times 1000 = 10,3 \sim 10 \text{ mmol}$$

c. Anilin (BM : 93,13 g/mol, berat jenis : 1,02 g/cm³)

Volume : 0,9 ml

$$\text{mmol anilin} : \frac{0,9 \times 1,02}{93,13} \times 1000 = 9,9 \sim 10 \text{ mmol}$$

II. Perhitungan persentase hasil sintesis berdasarkan mmol teoritis

Persentase hasil N-fenil-N'-benzoiltiourea :

	ammonium tiosianat	+	benzoil klorida	→	benzoilisotiosianat	+	NH ₄ Cl
awal	15 mmol		10 mmol		0		0
reaksi	10 mmol		10 mmol	-	10 mmol		10 mmol
sisa	5 mmol		0		10 mmol		10 mmol

	Benzoilisotiosianat	+	anilin	→	N-fenil-N'-benzoiltiourea
Awal	10 mmol		10 mmol		0
reaksi	10 mmol		10 mmol	-	10 mmol
sisa	0		0		10 mmol

Perhitungan persentase hasil sintesis :

BM N-fenil-N'-benzoiltiourea teoritis = 256,32

Sintesis I : Massa praktis = 2,11 gram

$$\text{Mmol praktis} = \frac{2,11}{256,32} \times 1000 = 8,23 \text{ mmol}$$

$$\text{Mmol teoritis} = 9,88 \text{ mmol}$$

$$\% \text{ hasil} = \frac{8,23}{9,88} \times 100 \% = 83,23 \% \approx 83 \%$$

Sintesis II : Massa praktis = 2,12 gram

$$\text{Mmol praktis} = \frac{2,12}{256,32} \times 1000 = 8,27 \text{ mmol}$$

$$\% \text{ hasil} = \frac{8,27}{9,88} \times 100 \% = 83,70 \% \approx 84 \%$$

Sintesis III : Massa praktis = 2,11 gram

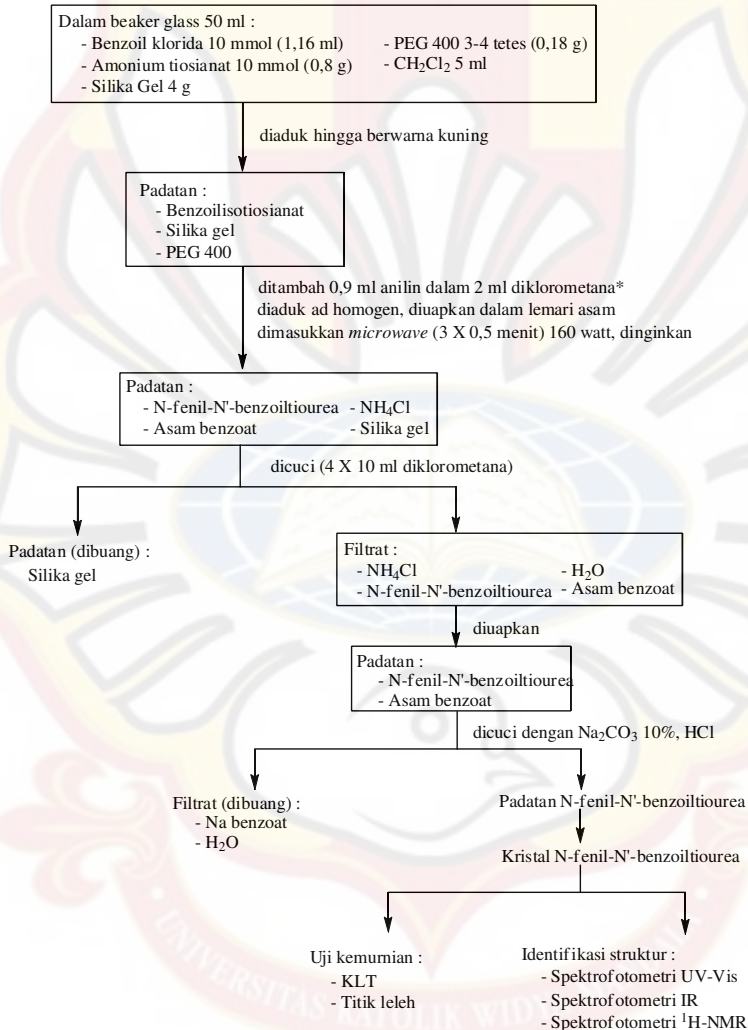
$$\text{Mmol praktis} = 8,23 \text{ mmol}$$

$$\% \text{ hasil} = \frac{8,23}{9,88} \times 100 \% = 83,23 \% \approx 83 \%$$

$$\text{Persentase hasil rata-rata : } \frac{83\% + 84\% + 83\%}{3} = 83,33 \%$$

LAMPIRAN B

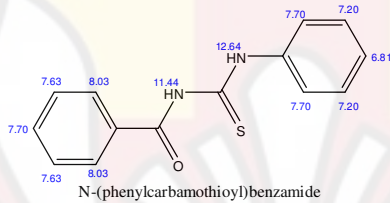
SKEMA KERJA SINTESIS SENYAWA TURUNAN N-FENIL-N'- BENZOILTIOUREA



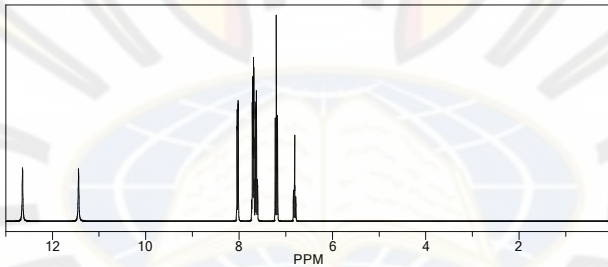
* pada sintesis N-(4-klorofenil)-N'-benzothiourea ditambah 1,27 gr 4-kloroanilin dalam 3 ml diklorometana

* pada sintesis N-(3,4-diklorofenil)-N'-benzothiourea ditambah 1,62 gr 3,4-dikloroanilin dalam 3 ml diklorometana

LAMPIRAN C

ESTIMASI RMI-¹H SENYAWA N-FENIL-N'-BENZOILTIOUREAChemNMR ¹H Estimation

Estimation quality is indicated by color: good, medium, rough

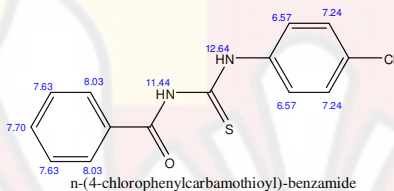
Protocol of the ¹H-NMR Prediction:

Node	Shift	Base + Inc.	Comment (ppm rel. to TMS)
NH	12.64	4.00	aromatic C-NH
		8.64	general corrections
NH	11.44	8.00	sec. amide
		3.44	general corrections
CH	7.70	7.26	1-benzene
		-0.80	1-N
		1.24	general corrections
CH	8.03	7.26	1-benzene
		0.69	1-C(=O)N
		0.08	general corrections
CH	7.70	7.26	1-benzene
		-0.80	1-N
		1.24	general corrections
CH	8.03	7.26	1-benzene
		0.69	1-C(=O)N
		0.08	general corrections
CH	7.20	7.26	1-benzene
		-0.25	1-N
		0.19	general corrections
CH	7.63	7.26	1-benzene
		0.18	1-C(=O)N
		0.19	general corrections
CH	7.20	7.26	1-benzene
		-0.25	1-N
		0.19	general corrections
CH	7.63	7.26	1-benzene
		0.18	1-C(=O)N
		0.19	general corrections
CH	6.81	7.26	1-benzene
		-0.64	1-N
		0.19	general corrections
CH	7.70	7.26	1-benzene
		0.25	1-C(=O)N
		0.19	general corrections

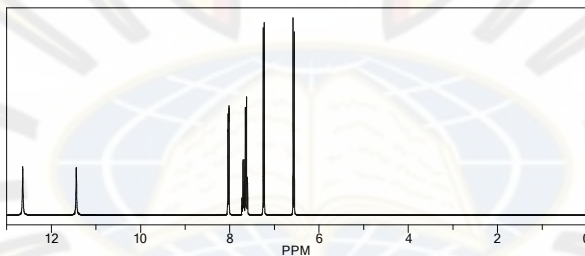
LAMPIRAN D

ESTIMASI RMI-¹H SENYAWA N-(4-KLOROFENIL)-N'- BENZOILTIOUREA

ChemNMR ¹H Estimation



Estimation quality is indicated by color: good, medium, rough



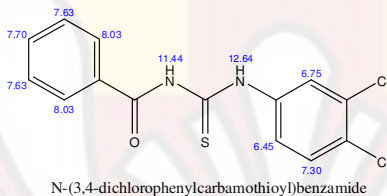
Protocol of the H-1 NMR Prediction:

Node	Shift	Base + Inc.	Comment (ppm rel. to TMS)
NH	12.64	4.00	aromatic C-NH
		8.64	general corrections
NH	11.44	8.00	sec. amide
		3.44	general corrections
CH	7.24	7.26	1-benzene
		-0.01	1 -Cl
		-0.25	1 -N
		0.22	general corrections
CH	6.57	7.26	1-benzene
		-0.06	1 -Cl
		-0.80	1 -N
		0.17	general corrections
CH	8.03	7.26	1-benzene
		0.69	1 -C(=O)N
		0.08	general corrections
CH	7.24	7.26	1-benzene
		-0.01	1 -Cl
		-0.25	1 -N
		0.22	general corrections
CH	6.57	7.26	1-benzene
		-0.06	1 -Cl
		-0.80	1 -N
		0.17	general corrections
CH	8.03	7.26	1-benzene
		0.69	1 -C(=O)N
		0.08	general corrections
CH	7.63	7.26	1-benzene
		0.18	1 -C(=O)N
		0.19	general corrections
CH	7.63	7.26	1-benzene
		0.18	1 -C(=O)N
		0.19	general corrections
CH	7.70	7.26	1-benzene
		0.25	1 -C(=O)N
		0.19	general corrections

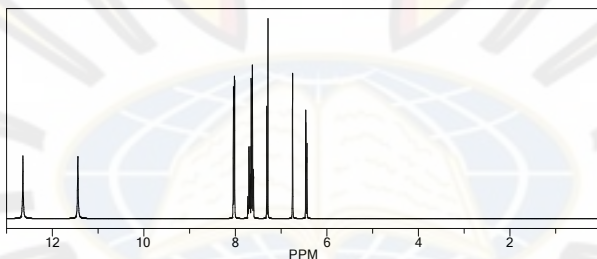
LAMPIRAN E

ESTIMASI RMI-¹H SENYAWA N-(3,4-DIKLOROFENIL)-N'- BENZOILTIOUREA

ChemNMR ¹H Estimation



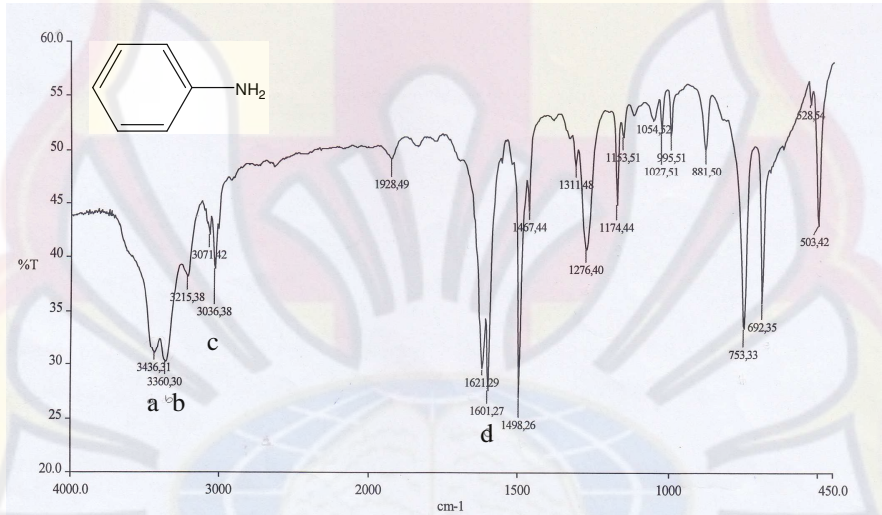
Estimation quality is indicated by color: **good**, **medium**, **rough**



Protocol of the H-1 NMR Prediction:

Node	Shift	Base + Inc.	Comment (ppm rel. to TMS)
NH	12.64	4.00	aromatic C-NH
		8.64	general corrections
NH	11.44	8.00	sec. amide
		3.44	general corrections
CH	6.75	7.26	1-benzene
		0.01	1 -Cl
		-0.06	1 -Cl
		-0.80	1 -N
		0.34	general corrections
CH	7.30	7.26	1-benzene
		-0.06	1 -Cl
		0.01	1 -Cl
		-0.25	1 -N
		0.34	general corrections
CH	6.45	7.26	1-benzene
		-0.12	1 -Cl
		-0.06	1 -Cl
		-0.80	1 -N
		0.17	general corrections
CH	8.03	7.26	1-benzene
		0.69	1 -C(=O)N
		0.08	general corrections
CH	8.03	7.26	1-benzene
		0.69	1 -C(=O)N
		0.08	general corrections
CH	7.63	7.26	1-benzene
		0.18	1 -C(=O)N
		0.19	general corrections
CH	7.63	7.26	1-benzene
		0.18	1 -C(=O)N
		0.19	general corrections
CH	7.70	7.26	1-benzene
		0.25	1 -C(=O)N
		0.19	general corrections

LAMPIRAN F
SPEKTRUM INFRAMERAH SENYAWA ANILIN DENGAN
MENGGUNAKAN PELLET KBR

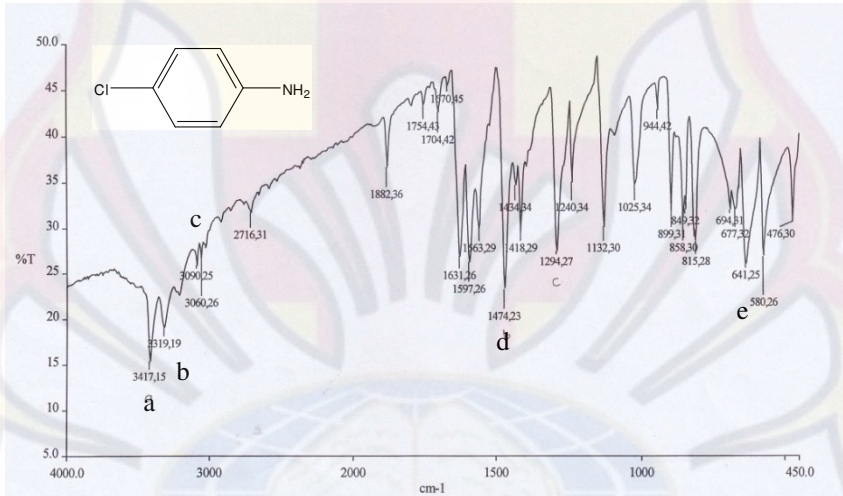


TABEL SERAPAN INFRAMERAH SENYAWA ANILIN

Ikatan	Bilangan gelombang (cm ⁻¹)		No. puncak
	Pustaka *	Anilin	
-NH ₂	3000 – 3700	3436, 3360	a, b
Csp ² -H	3000 – 3300	3036	c
C=C aromatis	1450 – 1600	1601	d

Keterangan : * Pavia *et al* 2001, Fessenden 1986, Hart 2003 , Williams 1966

LAMPIRAN G
SPEKTRUM INFRAMERAH SENYAWA 4-KLOROANILIN
DENGAN MENGGUNAKAN PELLET KBR

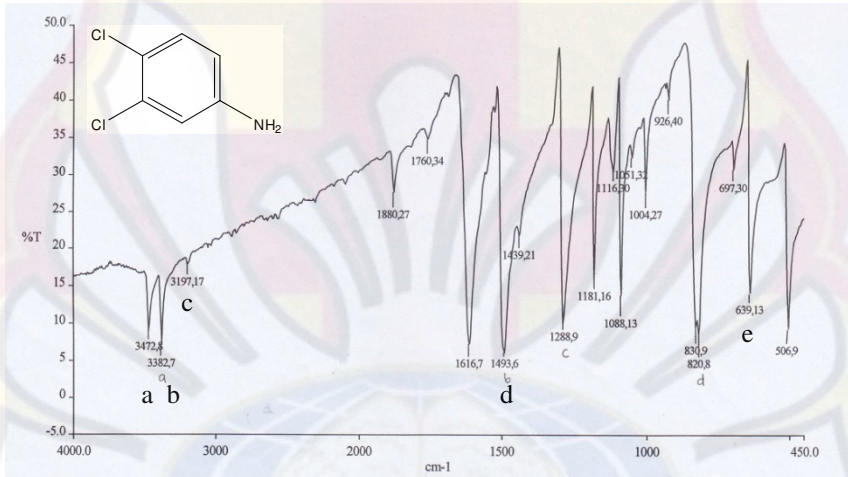


TABEL SERAPAN INFRAMERAH SENYAWA 4-KLOROANILIN

Ikatan	Bilangan gelombang (cm ⁻¹)		No. puncak
	Pustaka *	4-kloroanilin	
-NH ₂	3000 – 3700	3417, 3319	a, b
Csp ² -H	3000 – 3300	3090	c
C=C aromatis	1450 – 1600	1474	d
C-Cl	600 – 800	641	e

Keterangan : * Pavia *et al* 2001, Fessenden 1986, Hart 2003, Williams 1966.

LAMPIRAN H
SPEKTRUM INFRAMERAH SENYAWA 3,4-DIKLOROANILIN
DENGAN MENGGUNAKAN PELLET KBR



TABEL SERAPAN INFRAMERAH SENYAWA 3,4-DIKLOROANILIN

Ikatan	Bilangan gelombang (cm ⁻¹)		No. puncak
	Pustaka *	3,4-dikloroanilin	
-NH ₂	3000 – 3700	3472, 3382	a, b
Csp ² -H	3000 – 3300	3197	c
C=C aromatis	1450 – 1600	1493	d
C-Cl	600 – 800	639	e

Keterangan : * Pavia *et al* 2001, Fessenden 1986, Hart 2003, Williams 1966.