

## **BAB V**

### **KESIMPULAN DAN SARAN**

#### **V.1. Kesimpulan**

Kesimpulan yang didapat dari penelitian ini adalah:

1. Penambahan selulosa pada hidrogel cenderung meningkatkan tingkat kekerasan hidrogel.
2. Penambahan selulosa mampu meningkatkan *swelling ratio*, dimana uji *swelling ratio* untuk hidrogel GO/PVA/Selulosa yang tertinggi diperoleh dengan komposisi 1:1:0.4 dengan nilai 217% dari massa setelah pengeringan.
3. Penambahan selulosa dari ampas tebu mengurangi laju penyerapan hidrogel. Dimana laju penyerapan hidrogel tertinggi diperoleh dengan komposisi 1:1:0 (tanpa selulosa). Nilai  $Q_t$  pada waktu penyerapan 6 jam sebesar 84.28 mg/g pada suhu ruang.
4. Jenis adsorpsi untuk sistem ini adalah adsorpsi kimia karena nilai  $R^2$  yang terbaik diperoleh dengan menggunakan persamaan *pseudo-second order*.

#### **V.2. Saran**

Saran untuk penelitian berikutnya:

1. Diperlukan uji karakteristik fisik hidrogel secara kualitatif, seperti uji *tensile strenght*, *compressive strenght*.
2. Diperlukan studi lebih lanjut mengenai porositas hidrogel melalui uji SEM.
3. Diperlukan studi lebih lanjut mengenai pengaruh *graphene oxide* terhadap kapasitas adsorpsi *methylene blue* pada kondisi lainnya.

**DAFTAR PUSTAKA**

- [1] Kementerian Perindustrian Republik Indonesia, "Kementerian Perindustrian Republik Indonesia," 9 7 2017. [Online]. Available: <http://www.kemenperin.go.id/siaran-pers?&hal=5>.
- [2] Kementerian Perindustrian Republik Indonesia, "Populasi Sektor IKTA Meningkatkan, Tahun 2017 Ditargetkan 753 Perusahaan," in *Siaran Pers Kementerian Perindustrian Republik Indonesia*, Jakarta, 2017.
- [3] Y. Chen, L. Chen, H. Bai and L. Li, "Graphene oxide–chitosan composite hydrogels as broad spectrum adsorbents for water purification," *Journal of Material Chemistry A*, vol. 1, no. 6, p. 1992, 2013.
- [4] S. Babel and T. Kurniawan, "Low-cost adsorbents for heavy metals uptake from contaminated water: a review," *Journal of Hazardous Materials*, vol. 97, no. 1-3, pp. 219-243, 2003.
- [5] Y. Zhan, H. Zhang, J. Lin, Z. Zhang and J. Gao, "Role of zeolite's exchangeable cations in phosphate adsorption onto zirconium-modified zeolite," *Journal of Molecular Liquids*, vol. 243, pp. 624-637, 2017.
- [6] Z. Huang, Y. Li, W. Chen, J. Shi, N. Zhang, X. Wang, Z. Li, L. Gao and Y. Zhang, "Modified bentonite adsorption of organic pollutants of dye wastewater," *Materials Chemistry and Physics*, vol. 202, pp. 266 - 276, 2017.
- [7] H. Guo, T. Jiao, Q. Zhang, W. Guo, Q. Peng and X. Yan, "Preparation of Graphene Oxide-Based Hydrogels as Efficient Dye Adsorbents for Wastewater Treatment," *Nanoscale Research Letters*, vol. 10, p. 272, 2015.
- [8] M. Stoller, S. Park, Y. Zhu, J. An and R. Ruoff, "Graphene-Based Ultracapacitors," *Nanoletters*, vol. 8, no. 10, pp. 3498- 3502, 2008.

- [9] M. Allen, V. Tung and R. Kaner, "Honeycomb carbon: a review of graphene," *Chem Reviews*, vol. 110, no. 1, pp. 132-145, 2010.
- [10] C. Chang, A. Lue and L. Zhang, "Effects of Crosslinking Methods on Structure and Properties of Cellulose/PVA Hydrogels," *Macromolecular Chemistry and Physics*, vol. 209, pp. 1266-1273, 2008.
- [11] S. Liang, J. Wu, H. Tian, L. Zhang and J. Xu, "High-strength cellulose/poly(ethylene glycol) gels.," *ChemSusChem*, vol. 1, no. 6, pp. 558-563, 2008.
- [12] C. Chang, L. Zhang, J. Zhou, L. Zhang and J. Kennedy, "Structure and properties of hydrogels prepared from cellulose in NaOH/urea aqueous solution," *Carbohydrate Polymers*, vol. 82, pp. 122-127, 2010.
- [13] L. Thomas, U. Arun, S. Remya and P. Nair, "A biodegradable and biocompatible PVA-citric acid polyester with potential applications as matrix for vascular tissue engineering.," 2009.
- [14] X. Wei, T. Huang, Y. J.H., N. Zhang, Y. Wang and Z. Z.W., "Green Synthesis of Hybrid Graphene Oxide/Microcrystalline Cellulose Aerogels and Their use as superadsorbents," *Journal of Hazardous*, vol. 335, pp. 28-38, 2017.
- [15] A. Yakout, R. El-Sokkary, M. Shreadah and O. Hamid, "Cross-linked graphene oxide sheets via modified extracted cellulose with high metal adsorption," *Carbohydrate Polymers*, vol. 172, pp. 20-27, 2017.
- [16] Z. Liu, D. Li, H. Dai and H. Huang, "Enhanced properties of tea residue cellulose hydrogels by addition of graphene oxide," *Journal of Molecular Liquids*, vol. 244, pp. 110-116, 2017.
- [17] A. Kurniawan, H. Sutiono, Y. Ju, F. Soetaredjo, A. Ayucitra, A. Yudha and S. Ismadji, "Utilization of rarasaponin natural surfactant for organo-bentonite preparation: Application for methylene blue removal from aqueous effluent," *Microporous and Mesoporous Materials*, vol. 142, pp. 184-193, 2011.
- [18] W. Hidayat, *Teknologi Pengolahan Air Limbah*, Jakarta: Majari Magazine, 2008.

- [19] K. Rini, Dian. and A. Fendy, "Optimasi Aktivasi Zeolit untuk Dehumidifikasi," Undip, Semarang, 2010.
- [20] K. Soleimani, A. Tehrani and M. Adeli, "Bioconjugated graphene oxide hydrogel as an effective adsorbent for cationic dye removal," *Ecotoxicology and Environment Safety*, vol. 147, pp. 34-42, 2018.
- [21] X. Chen, S. Zhou, L. Zhang, T. You and F. Xu, "Adsorption of Heavy Metals by Graphene Oxide/Cellulose Hydrogel Prepared from NaOH/Urea Aqueous Solution," *Materials*, vol. 9, p. 582, 2016.
- [22] X. Yang, Y. Li, Q. Du, X. Wang, S. Hu, L. Chen, Z. Wang, Y. Xia and L. Xia, "Adsorption of Methylene Blue from Aqueous Solutions by Polyvinyl Alcohol/Graphene Oxide Composites," *Journal of Nanoscience and Nanotechnology*, vol. 16, pp. 1775-1782, 2016.
- [23] V. Chabot, D. Higgins, A. Yu, X. Xiao, Z. Chen and J. Zhang, "a review of graphene and graphene oxide sponge: material synthesis and applications to energy and the environment," *Energy & Environment Science*, vol. 7, no. 5, p. 1564, 2014.
- [24] D. Dreyer, S. Park, C. Bielawski and R. Ruoff, "the chemistry of graphene oxide," *Chemical Society Reviews*, vol. 39, pp. 228-240, 2009.
- [25] H. Poh, F. Sanek, A. Ambrosi, G. Zhao, Z. Sofer and M. Pumera, "Graphenes prepared by Staudenmaier, Hofmann and Hummers methods with consequent thermal exfoliation exhibit very different electrochemical properties," *Nanoscale*, vol. 4, p. 3515, 2012.
- [26] W. Hummer and R. Offeman, "Preparation of Graphene Oxide," 1958.
- [27] L. Zhang, Z. Wang, C. Xu, Y. Li, J. Gao, W. Wang and Y. Liu, "High strength graphene oxide/polyvinyl alcohol composite hydrogels," *Journal of Materials Chemistry*, vol. 21, p. 10399, 2011.

- [28] S. Stauffer and N. Peppas, "Poly(vinyl alcohol) hydrogels prepared by freezing-thawing cyclic processing," *POLYMER*, vol. 33, no. 18, p. 1992, 1991.
- [29] A. Setiawan, "Co-composting bagasse dengan sludge limbah industri menggunakan teknik aerasi dan pengaruhnya terhadap nilai/n," 2011.
- [30] A. Mandal and D. Chakrabarty, "Isolation of nanocellulose from waste sugarcane bagasse (SCB) and its characterization," *Carbohydrate Polymers*, vol. 86, pp. 1291-1299, 2011.
- [31] Arup Mandal, Debabrata Chakrabarty, "Isolation of nanocellulose from waste sugarcane bagasse (SCB) and its," 2011.
- [32] A. Chesson, "A laboratory method for the determination of fibre content of linen flax," *Journal of Agricultural Research*, vol. 22, pp. 43-46, 1979.
- [33] J. Duan, J. Jiang, J. Li, L. Liu, Y. Li and C. Guan, "The Preparation of a Highly Stretchable Cellulose Nanowhisker Nanocomposite Hydrogel," *Journal of Nanomaterials*, pp. 1-8, 2015.
- [34] K. Yuwawech, J. Wootthikanokkhan and S. Tanpichai, "Effects of Two Different Cellulose Nanofiber Types on Properties of Poly(vinyl alcohol) Composite Films," *Journal of Materials*, vol. 16, no. 1, 2015.
- [35] C. Zhou and Q. Wu, "A novel polyacrylamide nanocomposite hydrogel reinforced with natural chitosan nanofibers," *Colloids and Surfaces B: Biointerfaces*, vol. 84, pp. 155-162, 2011.
- [36] S. Tanpichai and K. Oksman, "Cross-linked nanocomposite hydrogels based on cellulose nanocrystals and PVA: Mechanical properties and creep recovery," *Composites part A: Applied Science and Manufacturing*, vol. 88, pp. 226-233, 2016.
- [37] T. Jayaramudua, K. Hyun-U, G. Xiaoyuan, Y. Lia, S. Kim and J. Kim, "Cellulose/polyvinyl alcohol based hydrogels for reconfigurable lens," vol. 9802, 2016.

- [38] X. Qi, X. Yao, S. Deng, T. Zhou and Q. Fu, "Water-induced shape memory effect of graphene," vol. 2, p. 2240, 2014.
- [39] C. H. T. A. M. J. A. Sellberg, "Ultrafast X-ray probing of water structure below the homogeneous ice nucleation temperature, Nature," vol. 510, pp. 381-384, 2014.
- [40] N. B. d. C. J. L. E. A. Jeremias de Souza Macedo, "Kinetic and calorimetric study of the adsorption of dyes on mesoporous," vol. 298, pp. 515-522, 2006.
- [41] I. Langmuir, "The Constitution And Fundamental Properties of Solids and Liquids Part I Solids," *Journal of The American Chemical Society*, vol. 38, no. 11, pp. 2221-2295, 1916.
- [42] H. Freundlich, "Of the adsorption of gases. Section II. Kinetics and energetics of gas adsorption. Introductory paper to section II," *Transactions of the Faraday Society*, vol. 28, pp. 195-201, 1932.
- [43] J. Chen, Y. Li, Z. Y. and Y. Zhu, "Preparation and characterization of graphene oxide reinforced PVA film with boric acid as a crosslinker," *Journal of applied Polymer science*, vol. 22, p. 132, 2015.
- [44] M. Dogan, M. Alkan, O. Demirbas, Y. Özdemir and C. Özmetin, "Adsorption kinetics of maxilon blue GRL onto sepiolite from aqueous solutions," *Chemistry Engineering Journal*, vol. 124, no. 1, pp. 89-101, 2006.